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Thermochemistry Thermochemistry deals with heat (energy) changes in chemical reactions. In chemical reactions heat is released or absorbed. If reaction absorbs heat then we call them endothermic

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When a balanced chemical equation not only indicates the quantities of the different reactants and products but also indicates the amount of heat evolved or absorbed, it is called thermochemical equation. Fractional coefficients may be used in writing a thermochemical equation. $H_2(g) + \frac{1}{2} O_2(g) \longrightarrow H_2O(l) + 285.8 \text{ KJ mol}^{-1}$

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- Writethe thermochemical equations for chemical reactions and other processes.
- Describehow energy is lost or gained during changes of state.
- Calculatedthe heat absorbed or released in a chemical reaction.

[Energy and Chemical Change - Glencoe](#)

Hess's law states that if two or more thermochemical equations can be added to produce a final equation for a reaction, then the sum of all the enthalpy changes for the individual reactions is the enthalpy change for the ... STUDY GUIDE. Y12 - Energetics 13 Terms. annabelsw6. Chapter 15 Chemical Thermodynamics 44 Terms. DavidSpivey. chemistry ...

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