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evidence-based approaches to address student needs, encouraged the use of effective techniques within a department or an institution, and addressed the challenges that arose along the way. The research-based strategies in "Reaching Students"

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Hint: add your answer to question
1 and four times your answer to
question 2. $4 + 4(7) = 4 + 28 =$

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32 4. Verify that there are 32 electrons pictured in the drawing of CCl_4 . Count each dot in the picture above: there are 24. There are also 8 more electrons represented by the 4 lines (each line represents two electrons). 5.

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14. Based on your answer to question 13, verify that this equation is true: This equation is called the “Combined Gas Law”. Notice that it contains all of the equations combined into one!

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Again, the pressure times the volume divided by the Kelvin temperature equals 1.18 for every trial in the table. 15.

[Information: Gas Pressure](#)

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Relating Electron Configurations to the Periodic Table In this section you will see how the periodic table serves as a road map for writing electron configurations. Get your periodic table out and get ready. Remember that a row on the

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(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO₂ are produced, so

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the number of moles of CO₂ that can be produced when 4 moles of propane combusts = $4(3) = 12$.

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