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Gay's law (more correctly referred to as Amontons's law) states that the pressure of a given mass of gas varies directly with the absolute temperature of the gas, when the volume is kept constant.. Mathematically, it can be written as: $P \propto T$. It is a special case of the ideal gas law.. Gay-Lussac is incorrectly recognized for the Pressure Law which established that the pressure of an enclosed gas ...

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